

Multicatalytic Synthesis of α -Pyrrolidinyl Ketones via a Tandem Palladium(II)/Indium(III)-Catalyzed Aminochlorocarbonylation/Friedel–Crafts Acylation Reaction

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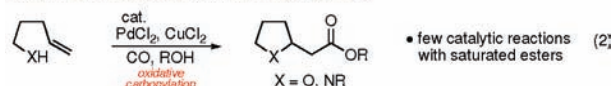
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As a manifestly practical science, synthetic organic chemistry counts as one of its primary objectives the invention of increasingly rapid, efficient, and economical strategies to access valuable molecular constructs. To this end, chemists have long been interested in tandem or cascade reactions, which deliver substantial increases in molecular complexity via single-pot operations without the need for intermediate workups or purifications.¹ Recently, there has been growing interest in the development of related *catalytic* multistep processes, whereby one or more catalysts promote two or more distinct chemical transformations in a single flask.^{2,3} Known as tandem-, cascade-, or sequential-catalysis, depending on the precise experimental details, such “multicatalytic” processes offer the potential to significantly advance the field of synthesis by helping to alleviate dependency on strictly iterative processes. This potential notwithstanding, the range of chemistries that have been exploited for multicatalytic purposes is quite narrow. With this in mind, our group is exploring the development of new multicatalytic processes with the broad goals of (1) developing highly efficient strategies for accessing privileged organic architectures and especially (2) inventing new chemical technologies that facilitate these processes (eq 1).



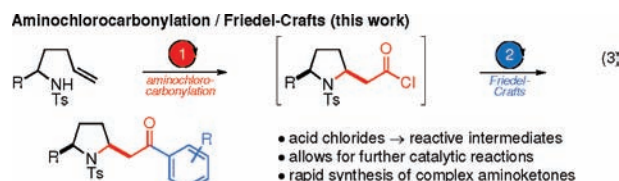
In this regard, we were attracted to the palladium(II) oxidative carbonylation chemistry developed by Semmelhack⁴ and Hegedus⁵ as an intriguing platform for multicatalytic design. Oxidative carbonylations are attractive transformations because they engender rapid increases in molecular complexity by the conversion of simple unsaturated alcohol or amine substrates to complex heterocyclic products (eq 2). Surprisingly, however, most reports of oxidative carbonylations detail termination by an alcohol nucleophile (usually as solvent) to generate carboxylic ester products.⁶

Oxidative carbonylation (Semmelhack, Hegedus)

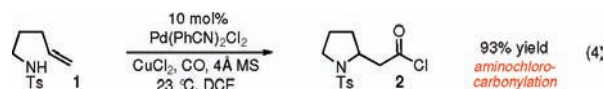


As useful as these transformations are, saturated esters are not efficient participants in many catalytic processes and thus do not readily lend themselves to multicatalytic methods. We were thus intrigued by the notion of diverting the oxidative carbonylation manifold to generate more reactive carbonyl products, which could then serve as viable intermediates for further catalytic operations. We report herein the finding that acid chlorides can be formed under oxidative carbonylation conditions and that these products participate in a tandem multicatalytic process involving Friedel–Crafts acylation (eq 3). Importantly, the products of this highly efficient

multicatalytic sequence are α -pyrrolidinyl ketones, a motif encountered in numerous biologically active natural isolates.⁷



Encouraged by the reports by Tsuji and Tkatchenko that PdCl₂ in the presence of CO can effect the transformation of alkenes to β -chloro acid chlorides,⁸ we reasoned that subjecting pentenyl amines to catalytic oxidative carbonylation conditions with the exclusion of alcohol should result in pyrrolidinyl acid chlorides. In fact, we found that when *N*-tosylpentenamine **1** was added slowly to a solution containing 10 mol % Pd(PhCN)₂Cl₂ and 3 equiv of CuCl₂ in dichloroethane (DCE) under a CO atmosphere, acid chloride **2** was isolated in 93% yield after filtration (eq 4). Notably, we saw only trace amounts of the amide adduct resulting from acylation of **1** by the acid chloride product **2**.



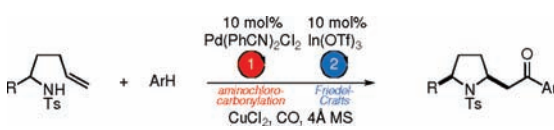
Next, in an effort to incorporate this aminochlorocarbonylation reaction into a useful multicatalytic process, we explored the possibility of employing the pyrrolidinyl acid chloride products for Lewis acid-catalyzed Friedel–Crafts acylation.⁹ A screen of Lewis acidic reagents revealed indium(III) triflate¹⁰ as a superior acylation catalyst in DCE at 23 °C. Importantly, the presence of indium(III) triflate and an electron-rich aromatic did not interfere with the aminochlorocarbonylation reaction. As such, we found it possible to execute a tandem aminochlorocarbonylation/Friedel–Crafts reaction by the slow addition of an amino substrate to a mixture of an aromatic nucleophile, 10 mol % Pd(PhCN)₂Cl₂, 10 mol % In(OTf)₃, CuCl₂, and 4 Å molecular sieves in DCE under a CO atmosphere. The β -pyrrolidinyl ketone products from this tandem process were isolated in good to high yields and with excellent diastereoselectivities (Table 1).

Importantly, a variety of electron-rich aromatic nucleophiles readily participate in this multicatalytic process, including aryl ether (entries 1–3), aryl bromide (entry 3), pyrrole (entry 4), thiophene (entry 5), and indole (entry 6) motifs.

This multicatalytic reaction also appears to be general in regard to substrate scope (Table 1). In addition to terminal olefinic substrates (entries 1–6), substrates bearing nonterminal olefins proved to be viable in this process, thus allowing the generation of products bearing substantial stereochemical complexity (entries 7

and 8). Surprisingly, in these cases the *E* and *Z* olefins give rise to the same product diastereomer, an observation that we attribute to thermodynamic equilibration via a retro-Mannich/Mannich addition process.¹¹ In addition, a substrate formally arising from a Mannich reaction was found to participate efficiently to produce the complex keto ester adduct shown in entry 9. Most interestingly, cyclic olefins were also ready participants in this tandem process, thus providing access to valuable fused or bridged bicyclic keto pyrrolidine products (entries 10 and 11).

Table 1. Multicatalytic Synthesis of Pyrrolidinyl Ketones^{a,b}

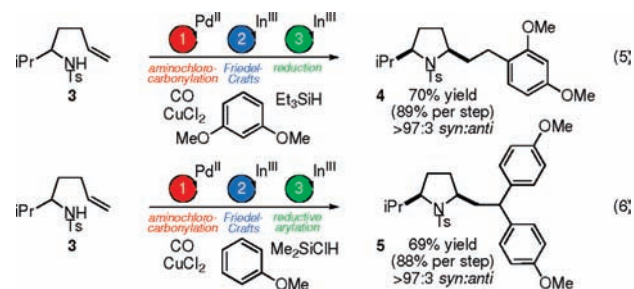


| Entry | Substrate / ArH | Product | % yield | d.r. |
|----------------|-----------------|---------|-----------------|-------|
| 1 | | | 85 | 96:4 |
| 2 | | | 92 | >97:3 |
| 3 | | | 66 | >97:3 |
| 4 | | | 88 ^c | >97:3 |
| 5 | | | 91 | >97:3 |
| 6 | | | 72 | >97:3 |
| <hr/> | | | | |
| 7 | | | 70 | 93:7 |
| 8 ^d | | | 58 | >97:3 |
| 9 | | | 74 | 95:5 |
| 10 | | | 85 | >97:3 |
| 11 | | | 75 | >97:3 |

^a Reactions were performed by syringe-pump addition of the substrate to a suspension of 10 mol % PdCl₂(PhCN)₂, CuCl₂ (3 equiv), 10 mol % In(OTf)₃, 4 Å molecular sieves, CO (1 atm), and aromatic nucleophile (5 or 10 equiv) in DCE at 23 °C. ^b Diastereomeric ratios were determined by ¹H NMR analysis of the crude reaction mixtures. ^c 3:1 regioselectivity. ^d 20 mol % In(OTf)₃ was used.

Finally, we have found it possible to merge this tandem reaction with additional catalytic transformations to realize three-step multicatalytic processes. For example, upon completion of the tandem aminochloro-carbonylation/Friedel–Crafts sequence with substrate **3** and dimethoxybenzene, addition of triethylsilane to the mixture effected an In(OTf)₃-catalyzed ketone reduction¹² to provide pyrrolidine adduct **4** in 70% yield (89% per step) with high

diastereoselectivity (eq 5). Alternatively, addition of chlorodimethylsilane precipitated a reductive arylation reaction to produce diaryl pyrrolidine **5** in an efficient 88% per step yield (eq 6).



In conclusion, we have developed the first catalytic aminochloro-carbonylation and have successfully incorporated this reaction into doubly and triply catalytic processes involving Friedel–Crafts acylation. We are currently exploring these and related multicatalytic processes.

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Supporting Information Available: Experimental procedures, product characterization data, and a CIF file. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- (1) Tietze, L. F.; Brasche, G.; Gericke, K. *Domino Reactions in Organic Synthesis*; Wiley-VCH: Weinheim, Germany, 2006.
- (2) For selected recent reviews on multicatalysis, see: (a) Ajamian, A.; Gleason, J. L. *Angew. Chem., Int. Ed.* **2004**, *43*, 3754. (b) Lee, J. M.; Na, Y.; Han, H.; Chang, S. *Chem. Soc. Rev.* **2004**, *33*, 302. (c) Wasilke, J.-C.; Obrey, S. J.; Baker, R. T.; Bazan, G. C. *Chem. Rev.* **2005**, *105*, 1001. (d) Enders, D.; Grondal, C.; Hüttl, M. R. M. *Angew. Chem., Int. Ed.* **2007**, *46*, 1570. (e) Chapman, C. J.; Frost, C. G. *Synthesis* **2007**, *1*. (f) Walji, A. M.; MacMillan, D. W. C. *Synlett* **2007**, 1477.
- (3) Kelly, B. D.; Allen, J. M.; Tundel, R. E.; Lambert, T. H. *Org. Lett.* Submitted for publication.
- (4) Semmelhack, M. F.; Zhask, A. *J. Am. Chem. Soc.* **1983**, *105*, 2034. (b) Semmelhack, M. F.; Bodurov, C. *J. Am. Chem. Soc.* **1984**, *106*, 1496. (c) Semmelhack, M. F.; Zhang, N. *J. Org. Chem.* **1989**, *54*, 4483. (d) Semmelhack, M. F.; Kim, C.; Zhang, N.; Bodurov, C.; Sanner, M.; Dobler, W.; Meier, M. *Pure Appl. Chem.* **1990**, *62*, 2035. (e) McCormick, M.; Monahan, M., III; Soria, J.; Goldsmith, D.; Liotta, D. *J. Org. Chem.* **1989**, *54*, 4485. (f) White, J. D.; Hong, J.; Robarge, L. A. *Tetrahedron Lett.* **1999**, *40*, 1463. (g) Holmes, C. P.; Bartlett, P. A. *J. Org. Chem.* **1989**, *54*, 98.
- (5) (a) Hegedus, L. S.; Allen, G. F.; Olsen, D. J. *J. Am. Chem. Soc.* **1980**, *102*, 3583. (b) Hegedus, L. S.; McKearin, J. M. *J. Am. Chem. Soc.* **1982**, *104*, 2444. (c) Hegedus, L. S.; Winton, P. M.; Varaprath, S. *J. Org. Chem.* **1981**, *46*, 2215.
- (6) (a) Beccalli, E. D.; Brogini, G.; Martinelli, M.; Sottocornola, S. *Chem. Rev.* **2007**, *107*, 5318. (b) Schmalz, H.-G.; Geis, O. In *Handbook of Organopalladium Chemistry for Organic Synthesis*; Negishi, E., Ed.; Wiley: New York, 2002; Vol. 2, p 2377.
- (7) A Beilstein substructure search indicated 1834 hits for natural products containing a pyrrolidinyl ring with β -oxygenation.
- (8) (a) Tsuji, J.; Morikawa, M.; Kiji, J. *Tetrahedron Lett.* **1963**, *4*, 1061. (b) Bonnet, M. C.; Carmona, N.; Tkatchenko, I. *J. Mol. Catal. A: Chem.* **1999**, *143*, 181.
- (9) Heaney, H. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon Press: New York, 1991; Vol. 3, p 733.
- (10) For use of In(OTf)₃ as an acylation catalyst, see: (a) Frost, C. G.; Hartley, J. P.; Griffin, D. *Tetrahedron Lett.* **2002**, *43*, 4789. (b) Chapman, C. J.; Frost, C. G.; Hartley, J. P.; Whittle, A. J. *Tetrahedron Lett.* **2001**, *42*, 773. (c) Chauhan, K. K.; Frost, C. G.; Love, I.; Waite, D. *Synlett* **1999**, 1743. Also see ref 12d.
- (11) Support for this hypothesis was garnered by the observation of 2',4'-dimethoxypropiophenone and 2-(2,4-dimethoxyphenyl)-*N*-tosylpyrrolidine in the crude reaction mixture.
- (12) (a) Yasuda, M.; Onishi, Y.; Ueba, M.; Miyai, T.; Baba, A. *J. Org. Chem.* **2001**, *66*, 7741. (b) Miyai, T.; Onishi, Y.; Baba, A. *Tetrahedron Lett.* **1998**, *39*, 6291. (c) Miyai, T.; Onishi, Y.; Baba, A. *Tetrahedron* **1999**, *55*, 1017. (d) Babu, S. A.; Yasuda, M.; Baba, A. *Org. Lett.* **2007**, *9*, 405.

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